PREDICTION OF TEMPERATURE RISE IN CONCRETE DUE TO HEAT OF HYDRATION OF CEMENT

Anura I.G.K. Mataraarachchi (09/8103)

Degree of Master of Philosophy

Department of Civil Engineering

University of Moratuwa

Sri Lanka

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Anura I.G.K. Mataraarachchi

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Thesis submitted in fulfillment of the requirements for the degree Master of
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Declaration

I declare that this is my own work and this thesis does not incorporate without acknowledgement any material previously submitted for a Degree or Diploma in any other University or institute of higher learning and to the best of my knowledge and belief it does not contain any material previously published or written by another person except where the acknowledgement is made in the text.

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Abstract

Temperature rise due to heat of hydration in concrete depends on many factors such as geometry of the concrete element, chemical, physical and thermal properties of concrete materials, mix proportion, initial temperature during concrete batching, and thermal boundary conditions during concrete hardening etc. The multicomponent cement hydration model developed by Maekawa et al., predicts the heat generation due to cement hydration based on cement contents, water contents, reference heat generation rate of main mineral components in cement, i.e. alite (C₃S), belite (C₂S), aluminate (C₃A), ferrite (C₄AF), and gypsum (CS₂H), mineral of cement, fineness of cement, thermal activity and interdependences of mineral components, and effects of consumption of free water during the hydration process etc. This cement hydration model was incorporated in the transient heat conduction analysis. The transient heat conduction analysis was carried out with ANSYS, finite element analysis software using Advance Parametric Design Language (APDL) computer programming to predict the temperature ruse due to heat of hydration of cement in concrete element for a given thermal boundary conditions.

Since the heat of hydration of cement is highly temperature dependent, variation of thermal properties of concrete at early ages is essential to predict the temperature response due to heat of hydration of cement in concrete. Experimental investigations were carried out to develop a model to estimate the variation of thermal conductivity of concrete from fresh to hardened state. The specific heat capacity of concrete (c) was estimated based on the specific heat capacities of cement powder and hydration products using Dulong – Petit Rule (DPR), Neumann – Kopp Rule (NKR), and mixing theory. Thermal conductivity of concrete (λ) was determined by fitting temperature rise curve at center of cube with temperature rise curve predicted by transient heat conduction analysis. Estimated specific heat capacity of concrete was applied in

transient heat conduction analysis program, to predict temperature rise curve from 1hrs to 1day for several mix proportions.

A mathematical model was developed to predict the variation of thermal conductivity based on experimentally investigated thermal conductivity data, mix proportions, thermal conductivity of concrete material found in literature, cement and water contents, formation, shapes, and saturation of gel and capillary pores of cement paste, degrees of hydration, surface saturation of aggregates by applying into general and effective medium theories used in estimation of effective thermal conductivity of a multicomponent material. The developed model was calibrated and verified with experimental data of concrete cube samples for several mix proportions. A computer program was developed using APDL coding of ANSYS software to predict the thermal properties of concrete once mix proportion, chemical, physical and thermal properties of concrete materials are known. This model was coupled with the multicomponent heat of hydration model to improve the program's ability to predict temperature rise with effects of variation of thermal properties with degree of hydration of cement.

The developed multicomponent heat of hydration model was calibrated and verified with temperature rise data detained from several field tests which were carried out in several construction projects in Sri Lanka. Measured and predicted temperature response are in good agreement, and therefore the proposed model can be used to predict temperature rise when chemical composition, mix proportions, and thermal boundary conditions are known.

Furthermore, the developed hydration model was used to obtain appropriate values for T1 (i.e. temperature drop between hydration peak and ambient temperature under local conditions which are required in design of water retaining structures.

Keywords: heat of hydration, thermal conductivity, specific heat capacity, early age concrete, transient heat conduction analysis.

This thesis is dedicated

To my Parents, Loving Wife and Son

Without whom none of my success would be possible

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List of Symbols

Latin Symbols

OPC Ordinary Portland Cement

APDL Advance Parametric Design Language

THW Transient Hot Wire
TLS Transient Line Source
THS Transient Hot Strip
TPS Transient Plane Source

C Carbon
Mg Magnesium
Al Aluminum
Fe Ferrous
Si Silicon
Ca Calcium

 C_3S 3CaO.SiO₂ - Alite C_2S 2CaO.SiO₂ - Belite

C₃A 3CaO.Al₂O₃ - Aluminate C₄AF 4CaO.Al₂O₃.Fe₂O₃ - Ferrite CSH Calcium Silicate Hydrates

CS2H Gypsum
AEt & FEt Ettringite

CH Calcium Hydroxide

LH Low Heat

SR Sulfate Resistant
DPR Dulong-Petit Rule
NKR Neumann - Kopp Rule

HS Hashin-Strikman Bound Theorem

MM Maxwell Model

 $\begin{array}{lll} \text{MEL} & \text{Maxwell} - \text{Eucken Limits} \\ \text{EMT} & \text{Effective Medium Theory} \\ \text{DoH} & \text{Degrees of Hydration} \\ \text{T}_0 & \text{Temperature at 0 K} \\ \text{T}_{100} & \text{Temperature at 100 K} \\ \end{array}$

Q_R Volumetric Heat Generation Rate

K Thermal Conductivity

 K_0 Thermal Conductivity of Concrete at 0 K K_{100} Thermal Conductivity of Concrete at 100 K

M Fractional Mass
X Molar Fraction

P Pressure
V Volume
W Weight
A Area
Q Heat Flow

 T_s Surface Temperature T_a Ambient Temperature

NLS Number of Time Intervals

H_c Specific Heat Generation Rate for Cement Powder

p Fractional Mass of Cement Minerals

Qi Accumulated Heat
R Gas Constant
E/R Thermal Activity
T Temperature

si Fineness of Cement PowderSi Blaine Value of Cement Minerals

S_{io} Reference Blaine Value of Cement Minerals

WC Unite Weight of Cement

P3A Weight Percentage of Aluminate
P3S Weight Percentage of Alite
P4AF Weight Percentage of Ferrite
P2S Weight Percentage of Belite
PPCS2H Weight Percentage of Gypsum
PPC Weight Percentage of OPC

BLN Blaine Value of OPC

PSG Weight Percentage of Slab
BLNSG Blaine Value of Slag

SGCS2H Weight Percentage of Sulfate in Slag

PFA Weight Percentage of Fly Ash
BLNFA Blaine Value of Fly Ash

FACS2H Weight Percentage of Sulfate in Fly Ash PLS Weight Percentage of Superplasticizer

BLNLS Blaine Value of Superplasticizer

WP Water to Powder Ratio

QSP Constant for the Effects of Superplasticizer Dosage
QSPAD Constant for the Effects of Superplasticizer Dosage
CHARSP Constant for the Effects of Superplasticizer Dosage

QSGMX Final Accumulated Heat Generated by Slag

RSGW1 Weight Percentage of Consumed Water of by Slab

RSGCA Weight Percentage of Consumed Calcium

Hydroxide when reacts with Slag

QFAMX Final Accumulated Heat Generated by Fly Ash
RFAW1 Weight Percentage of Consumed Water by Fly Ash

RFACA Weight Percentage of Consumed Calcium

Hydroxide when reacts with Fly Ash

RH3AMN Factor for Mono-sulfate Conversion

w Velocity of Wind

h Convection Coefficient

E_N Solar Radiation on Horizontal Surface E_V Solar Radiation on Vertical Surface

 $\begin{array}{ccc} L_a & & Latitude \\ L_o & & Longitude \end{array}$

LST Local Standard Time

LSM Local Standard Time Meridian

 $\begin{array}{cc} q & & \text{Heat Flux} \\ T_{dp} & & \text{Dew Point} \end{array}$

Greek Symbols

c Specific Heat Capacity of Concrete

ρ Density of Concrete

Retarding effects of fly ash and admixture

β Free water content

μ Effects of weight percentage of clinker minerals
 λ Effects of calcium hydroxide production on reaction

of fly and slag

Ø Porosity

δ Solar Declination Angle σ Stefan – Boltzmann Constant

ε Emissivity

α Thermal Diffusivity

Subscript

i, j, ktx, y, zCountersTimeDirections