A THREE DIMENSIONAL COMPUTATIONAL FLUID DYNAMICS MODEL FOR PYROLYSIS OF THERMALLY THICK BIOMASS PARTICLES

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Degree of Master of Science

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Declaration

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Abstract

The solid biomass especially the wood has been used as a source of energy for centuries in the world. However, the present world has started giving it a more value not because it is known for a longer time. But as a renewable energy source it lessens the fossil fuel depletion and be a part of climate change mitigation. Conversion of biomass to energy is done through several methods such as bio-chemical conversion and thermo-chemical conversion. In the present work, the drying and pyrolysis process of a thermally thick single wood particle has been investigated. A novel approach has been introduced considering the two-phase gas and solid inside the particle are not in thermal equilibrium. Mathematical relationship was built to determine distinct temperatures at the boundaries of solid and gas. An unsteady threedimensional (3D) model is developed and simulated in Computational Fluid Dynamics (CFD) framework. The Euler-Euler approach for modeling of single biomass particle has been succeeded with the help of C++ CFD toolbox in OpenFOAM. The 3D model can simulate the thermochemical conversion process of different particle types, particularly for different shapes to examine the spatial variations during the process. The model was validated by comparing the simulation results with data obtained by experiments conducted using a single particle reactor. Further, the model was applied in torrefaction of single wood particle then expanded to thermochemical conversion in packed bed.

Key words: Biomass CFD Mathematical model Thermally thick particle Thermochemical conversion Dedicated to my mother and farther for their unconditional love, endless support and encouragement.

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LIST OF ABBREVIATIONS

а	Absorption coefficient (m ⁻¹)
a _i	Fraction of component
A	Total specific area of solid phase (m ⁻¹)
A _i	Pre-exponential factor (s ⁻¹)
A_p	Particle surface area (m ²)
A _{spc}	Specific surface area of char (m ⁻¹)
A_w	Pre-exponential factor for water evaporation (s ⁻¹)
С	Linear anisotropic phase function coefficient
C _{eff}	Effective specific heat capacity of the solid $(J kg^{-1}K^{-1})$
<i>C</i> _{<i>p</i>,<i>g</i>}	Specific heat capacity of gas (J kg ⁻¹ K ⁻¹)
<i>C</i> _{<i>p,w</i>}	Specific heat capacity of water (J kg ⁻¹ K ⁻¹)
d	Particle diameter (m)
D _{<i>i</i>,<i>g</i>}	Gas diffusion coefficient (m ² s ⁻¹)
E _i	Activation energy (J mol ⁻¹)
E _w	Activation energy for water evaporation (J mol ⁻¹)
FSP	Fibre saturation point
G	Incident radiation (W m ⁻²)
h _{rad}	Radiative heat transfer coefficient (W $m^{-2} K^{-1}$)
k_g	Thermal conductivity of gas phase (W m ⁻¹ K ⁻¹)
$k_{m,i}$	Mass transfer coefficient of species (m s ⁻¹)
L	Characteristic length (m)

LHV _{feed}	Lower heating value of feed	
LHV _{torrefied}	Lower heating value of torrefied material	
M _{air}	Molecular weight of air (kg mol ⁻¹)	
M _{char}	Char molecular weight (kg mol ⁻¹)	
M _{feed}	Mass of feed (kg)	
M _i	Amount of species (kg m ⁻³)	
M _s	Molecular weight of specie (kg mol ⁻¹)	
M _{solid}	Amount of solid species (kg m ⁻³)	
M _{torrefied}	Mass of torrefied material (kg)	
n	Refractive index of the medium	
p	Gas pressure (Pa)	
$Q_{charconversion}$ Char conversion source term		
\boldsymbol{Q}_{drying}	Drying heat (J m ⁻³ s ⁻¹)	
$Q_{gasgeneration}$ Total heat interaction during solid to gas conversion (J m ⁻³ s ⁻¹)		
$Q_{pyrolysis}$	Pyrolysis heat (J m ⁻³ s ⁻¹)	
Q_r	Radiation flux	
$Q_{radiation}$	Radiation source term	
R	Universal gas constant (J K ⁻¹ mol ⁻¹)	

R _{char,i}	Char reaction rate (kg $m^{-3} s^{-1}$)
R _{char,k,i}	Kinetic rate of char reactions (kg m ⁻³ s ⁻¹)
R _{char,m,i}	Mass transfer rate of char reactions (kg m ⁻³ s ⁻¹)
R _{drying}	Rate of drying (kg m ⁻³ s ⁻¹)
R _i	Pyrolysis rate of <i>i</i> component (kg m ⁻³ s ⁻¹)
R _{pyrolysis}	Total rate of pyrolysis (kg m ⁻³ s ⁻¹)
S _{drying}	Drying source term
S _{pyrolysis}	Pyrolysis source term
S _{charconversio}	_n Char conversion source term
t	Time (s)
T _{b,eff}	Effective boundary temperature (K)
$T_{b,s}$	Solid boundary temperature (K)
T _{evp}	Boiling pint of water (K)
T_g	Gas phase temperature (K)
T _i	Initial temperature (K)
T _s	Solid phase temperature (K)
T _{sg}	Surrounding gas temperature (K)
T _w	Reactor wall temperature (K)
u_g	Gas phase velocity (m s ⁻¹)
v_i	Stoichiometric coefficient of specie
Y _b	Fraction of bound water

Y _i	Mass fraction of gas species
Y _w	Mass fraction of wood species
Δh_{bw}	Evaporation heat of bound water (J kg ⁻¹)
$\Delta \boldsymbol{h}_{char,i}$	Char reaction heat (J kg ⁻¹)
Δh_{fw}	Evaporation heat of free water (J kg ⁻¹)
$\Delta h_{pyrolysis}$	Heat of pyrolysis (J kg ⁻¹)
Δt	Time step (s)
$\boldsymbol{\varepsilon}_{g}$	Volume fraction of gas porosity
<i>E</i> _s	Volume fraction of solid porosity
κ _c	Char permeability
ĸ _w	Wood permeability
ε	Wall emissivity
ρ_c	Mass of char (kg m ⁻³)
ρ _{df}	Density of dry wood (kg m ⁻³)
$ ho_g$	Gas density (kg m ⁻³)
$\rho_{H_20,b}$	Bound water in the wood particle (kg m ⁻³)
$\rho_{H_20,f}$	Free water in the wood particle (kg m ⁻³)
ρ_i	Density of specie (kg m ⁻³)
ρ_{is}	Density of specie at the reacting surface (kg m ⁻³)
$ ho_w$	Density of moisture (kg m ⁻³)
μ_g	Dynamic viscosity (Ns m ⁻²)
σ	Stefan-Boltzmann constant (Wm K ⁻⁴)

- σ_i Average collision diameter (\dot{A})
- σ_s Scattering coefficient (m⁻¹)
- Ω_i Diffusion collision integral
- $\boldsymbol{\varOmega}_{\boldsymbol{v}}$ Stoichiometric coefficient
- δt Time step (s)
- |*U*| Magnitude of the velocity through a particular cell
- δx Cell size in the direction of velocity